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tert-Butyldimethylsilyldihalomethyllithium as a Dihalomethylene Dianion Synthon. 1,3-Rearrangement and 1,4-Rearrangement of Silyl Group from Carbon to Oxide

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Abstract: One-pot synthesis of $R^1CH(OSiMe_2-t\text{-Bu})CX_2CH(OH)R^2$ (X=Cl, Br) by successive addition of two different aldehydes (R^1CHO and R^2CHO) has been achieved starting from tert-butyldimethylsilyldihalomethyllithium. Treatment of a THF solution of the title carbanion (X=Cl) with $p\text{-MeOC}_6H_4CHO$ or n-BuCHO followed by an addition of HMPA and benzaldehyde gave the corresponding 1,3-diol monosilyl ether in 83% or 45% yield, respectively. The use of oxiranc in place of aldehyde as the first electrophile followed by addition of benzaldehyde provided 1,4-diol monosilyl ether.

(1) Reaction of text-butyldimethylsilyldihalomethyllithium with aldehydes followed by 1,3-rearrangement of silvl group from carbon to oxide.

Intramolecular 1,2-rearrangement of silicon from carbon to negatively charged oxygen is well known as Brook rearrangement 1 and many examples have been reported 2 for the construction of organic molecules. In contrast, 1,3-rearrangement of silicon from carbon to β -oxyanion is rare since olefin formation via 1,2-climination of β -oxidosilanes is rapid. We wish to report a synthetic method for formation of two carbon-carbon bonds in one-pot based on organosilicon chemistry which involves an unprecedented 1,3-rearrangement of silicon 3 , 4, 5

tert-Butyl(dibromomethyl)dimethylsilane (1a)⁶ was deprotonated by treatment with lithium disopropylamide in DME-THF (2:1) at -78 °C to give tert-butyldimethylsilyldibromomethyllithium (2a). Treatment of 2a with benzaldehyde (2.4 eq) lead to 1,3-diol monosilyl ether 5 (1:2 adduct, 72% yield) via the intermediacy of lithium carbenoid 4 along with 1:1 adduct (PhCH(OSiMe₂-t-Bu)CHBr₂, 6, 22%). This was

Scheme 3

a surprising result since the intermediate β -oxidosilane 3, by analogy with the examples of Me₃SiCH(Li)Cl⁷ and Me₃SiC(Li)(SR)₂,⁸ would have been expected to eliminate lithium bromide or lithium tert-butyldimethylsilanoxide to give α,β -epoxy silane or alkene rather than 1,3-diol monosilyl ether 5 (Scheme 1).

The distribution of the products (1:1 adduct to 1:2 adduct) depends heavily on the nature of the substituent on the silicon. The respective dibromomethylsilane and the yields⁹ of the corresponding products (1:1 adduct and 1:2 adduct) in the reaction of R₃SiCLiBr₂ (1.2 mmol) with PhCHO (1.0 mmol) in THF were as follows: Me₃SiCHBr₂, ¹⁰ 29%, 0%; t-BuMe₂SiCHBr₂, 68%, 22%; ¹¹ i-Pr₃SiCHBr₂, 18%, 25%; Ph2MeSiCHBr2, 36%, 49%; Ph3SiCHBr2, 18%, 74%. Thus, Ph3SiCLiBr2 was the best reagent for the preparation of 1,3-diol monosilyl ether (PhCH(OSiR₃)CBr₂CH(OH)Ph). 12 The rate of rearrangement was also sensitive to the reaction solvent. In ether, instead of DME-THF, rearrangement of silicon $(3 \rightarrow 4)$ did not proceed and the reaction of tert-butyldimethylsilyldibromomethyllithium (2a) with benzaldehyde gave an adduct PhCH(OH)CBr2(SiMe2-t-Bu), 7 in 77% yield after work up (1 N HCl-ether). Addition of methanol (10 eq) before workup to the reaction mixture provided the rearranged product 6 in 87% yield. In the same way, the reaction between 2a and heptanal, cinnamaldehyde, or acetophenone provided the corresponding rearranged silyl ether R ¹R²C(OSiMe₂-t-Bu)CHBr₂ in 71%, 75% or 65% yield, respectively, by the addition of methanol before workup. Rearrangement by an addition of methanol might proceed as follows: (1) Protonation of 3 by methanol gives 7 and lithium methoxide, (2) lithium methoxide can deprotonate 7 to regenerate 3 and an equilibrium mixture of 3 and lithium methoxide is obtained, (3) equilibration shifts via C→O rearrangement of silyl group to form dibromoalkyllithium 4, and (4) finally protonation of 4 by methanol affords the rearranged product 6 (Scheme 2). This assumption was supported by the following two facts. The use of MeOD gave PhCH(OSiMe₂-t-Bu)CDBr₂. When the carbinol 7 (0.5 mmol) was treated with a catalytic amount of CH₃OLi (0.1 mmol) in ether (3 ml)-methanol (5.0 mmol), the carbinol was transformed rapidly to the alkoxysilane 6 in 90% yield.

Scheme 2

OLi

$$Ph \longrightarrow SiMe_2 t \cdot Bu$$
 $Br \quad 3$
 $OSiMe_2 t \cdot Bu$
 $Ph \longrightarrow SiMe_2 t \cdot Bu$
 $OSiMe_2 t \cdot B$

A crossover experiment was conducted to demonstrate the intramolecularity of the migration process. Upon treatment of a mixture of 7 and 8 with a catalytic amount of CH₃OLi in Et₂O-MeOH at -78 °C for 1 h, only two products (6 and 9) were isolated. No crossover products could be observed (Scheme 3).

Ph
$$\xrightarrow{Si}$$
 + $\xrightarrow{n-Bu}$ $\xrightarrow{Sii-Pr_3}$ $\xrightarrow{Et_2O / CH_3OH}$ \xrightarrow{Ph} \xrightarrow{Br} $\xrightarrow{B$

Treatment of 6 with lithium disopropylamide in THF provided carbanion 4 which reacted with electrophile such as methyl iodide, allyl bromide, benzaldehyde, pentanal, or cyclohexanone to give the corresponding adduct in 97% (10a), 97% (10b), 95% (5), 85% (10c), or 70% (10d) yield, respectively (Scheme 4).

Scheme 4

Then we turned out our attention toward one-pot synthesis of 10 by successive addition of two different electrophiles to *tert*-butyldimethylsilyldibromomethyllithium (2a). It was anticipated that an addition of DME and second electrophile to the reaction mixture of 2a and benzaldehyde in ether would provide 10 in one-pot. However, an addition of DME and methyl iodide or 4-methoxybenzaldehyde as a second electrophile gave no desired product and only 6 was isolated in 50–55% yield. An addition of HMPA instead of DME afforded an adduct 10 (E'=Me) in 53% yield upon successive treatment with Mel as the second electrophile. Fortunately, *tert*-butyldimethylsilyldichloromethyllithium (2b), generated from *tert*-butyl(dichloromethyl)dimethylsilane (1b) and LDA, proved to be more effective than dibromo analogue 2a for the purpose. In this case, the migration of silicon in the adduct 11, derived from 2b and aldehyde such as PhCHO, PhCH=CHCHO, or *n*-BuCHO, did not proceed in THF. An addition of HMPA to the reaction mixture, however, caused the rearrangement providing carbon anion which reacted with various second electrophiles effectively (Table 1).

Table 1. One-pot synthesis of RCH(OSiMe₂-t-Bu)CCl₂E' from tert-butyl(dichloromethyl)-dimethylsilane (1b)

t-BuMe₂S 1b	iiCHCl ₂ 1) LDA 2) RCHO -78 °C	OLi SiMe ₂ t-Bu	1) Electrophile HMPA → r. t. or −20 °C 2) H ₃ O 9	OSiMe ₂ t-Bu
	R	Electrophile	E,	Yield of 12 (%)
а	Ph	Mel	Me	71
b	Ph	CH2=CHCH2Br	CH ₂ =CHCH ₂	7 0
c	PhCH=CH	Mel	Me	74
d	n-Pr	CH2=CHCH2Br	CH ₂ =CHCH ₂	40
e	4-MeO-C ₆ H ₄	PhCHO	PhCH(OH)	83a
f	PhCH=CH	PhCHO	PhCH(OH)	73a
g	n-Bu	PhCHO	PhCH(OH)	45a

a) The products consist of two monosilyl ethers such as $PhCH(OH)CCl_2CH(OSiMe_2-t-Bu)C_6H_4-p-OMe$ and $PhCH(OSiMe_2-t-Bu)CCl_2CH(OH)C_6H_4-p-OMe$. Each isomer was a mixture of two diastereomers ((1R*,3R*):(1R*,3S*)=4:6 or 1:1).

Dichlorides were easily reduced by $n\text{-Bu}_3\text{SnH-Et}_3\text{B}^{13}$ to give the corresponding methylene compounds. For instance, treatment of 12a (0.6 mmol) with $n\text{-Bu}_3\text{SnH}$ (1.75 mmol) in the presence of Et_3B (0.7 mmol) in hexane at 80 °C afforded PhCH(OSiMe₂-t-Bu)CH₂CH₃ in 97% yield which was converted into 1-phenyl-1-propanol (13) by treatment with $n\text{-Bu}_4\text{NF}$. Thus, tert-butyldimethylsilyldichloromethyllithium can be regarded as a synthon of dichloromethylene dianion (CCl₂²⁻) or methylene dianion (CH₂²⁻) (Scheme 5). 14

(2) Reaction of text-butyldimethylsilyldihalomethyllithium with oxiranes followed by 1,4-rearrangement of silvl group from carbon to oxide.

The new method described in section (1) was applied to the reaction with oxiranes. Treatment of 2-phenyloxirane (14a) with *tert*-butyldimethylsilyldibromomethyllithium (2a) in ether at -40 °C¹⁵ provided 3,3-dibromo-3-*tert*-butyldimethylsilyl-1-phenyl-1-propanol (16a) in 32% yield. Other oxiranes such as 14b or 14c also gave the corresponding alcohol 16b or 16c in 51% or 80% yield, respectively (Scheme 6). The reaction did not proceed at -78 °C in contrast to the reaction with aldehyde which took place easily at that temperature. Di-substituted oxiranes such as 1,2-epoxycyclopentane and 2-methoxymethyl-3-phenyloxirane did not react with 2a and oxiranes were recovered unchanged even after prolonged reaction period. 2-Phenyloxetane and 2-methoxymethyloxolane were also recovered upon treatment with 2a.

Scheme 6

t-BuMe₂SiCBr₂

2a Li

$$Si = t$$
-BuMe₂Si

 $Si = t$ -BuMe₂Si

a: R = Ph b: R = CH₃ C: R = CH₂OCH₃

Then we studied the 1,4-rearrangement ¹⁶ of silyl group from carbon to oxide in the adduct **15** and found that the rate of the rearrangement depended heavily on the reaction solvent as in the case of the adduct **3** generated from **2a** and aldehyde. In ether, migration did not take place. However, change of the solvent from ether to THF facilitated the 1,4-rearrangement of silyl group. ¹⁷ For instance, treatment of 1,2-epoxypropane with **2a** in THF at -40 °C gave 1,1-dibromo-1-deuterio-3-tert-butyldimethylsiloxybutane in 83% yield (81% D) after quenching with MeOD. Various oxiranes provided the corresponding products as shown in Table 2. Among them, ethylene oxide gave the best results and the reaction with **2a** afforded 3,3-dibromo-1-siloxypropane almost quantitatively. t-Butyldimethylsilyldichloromethyllithium (**2b**) reacted with oxiranes equally effectively as **2a**.

Table 2. Reaction of tert-butyldimethylsilyldihalomethyllithium 2 with oxiranes in THF

a) MeOD was used instead of MeOH

Dihaloalkyllithium 18, regenerated by 1,4-rearrangement of silyl group in THF in the presence of HMPA smoothly reacted with second electrophiles to give the corresponding adducts in good yields. The representative results are summarized in Table 3. The use of isopropyl formate afforded 2,2-dichloro-4-siloxybutanal.

Table 3. One-pot synthesis of RCH(OSiMe2-1-Bu)CH2CX2E' from 2

Si CLi
$$X_2$$
 THF R OLi X X HMPA R OSi X X Electrophile \rightarrow r. t. or \rightarrow

	X	R	Electrophile	E'	Yield of 19 (%)
a	Br	CH ₃	CH ₃ I	CH ₃	60
b	Cl	CH ₃	CH ₃ I	CH ₃	68
c	Cl	CH ₃	PhCHO	PhCH(OH)	65
d	Cl	Н	CH ₃ I	CH ₃	80
e	Cl	Н	HCOO <i>i</i> Pr	CHO	56

Finally, we examined the relative reaction rate between 1,3-rearrangement and 1,4-rearrangement. A catalytic amount of *tert*-BuOK was added to a mixture of 7 and 16a (7:16a = 1:1) in CD₃OD. The reaction mixture was monitored by ¹H NMR (PhCH vs PhCH(OSi)). Whereas 1,3-rearrangement completed within 5 min, 1,4-rearrangement was slow and took 30 min to complete (Scheme 7).

Scheme 7

Ph
$$Si$$
 + Ph Si CD_3OD Ph D + Ph D Ph D + Ph D Ph D

Experimental

Distillation of the products was performed by the use of Kugelrohr (Büchi), and boiling points are indicated by air-bath temperature without correction. Melting points were obtained on a Yanako MP-50929 melting point apparatus and are uncorrected. ^{1}H NMR and ^{13}C NMR spectra were taken on a Varian GEMINI 300 spectrometer, CDCl3 was used as a solvent, and chemical shifts being given in δ with tetramethylsilane as an internal standard. IR spectra were determined on a JASCO IR-810 spectrometer. The analyses were carried out at the Elemental Analysis Center of Kyoto University. Toluene, hexane, and diethyl ether were dried over a slice of sodium. Tetrahydrofuran (THF) was freshly distilled from sodium benzophenone ketyl before use.

tert-Butyl(dibromomethyl)dimethylsilane (1a): Bp 60 °C (1 Torr); IR (neat) 2926, 2856, 1464, 1364, 1252, 839, 824, 779 cm $^{-1}$; 1 H NMR (CDCl $_{3}$) δ 0.25 (s, 6H), 1.02 (s, 9H), 5.27 (s, 1H); 13 C NMR (CDCl $_{3}$) δ -6.84, 17.94, 27.30, 34.11. Found: C, 29.22; H, 5.76%. Calcd for C $_{7}$ H $_{16}$ Br $_{2}$ Si: C, 29.18; H, 5.60%.

tert-Butyl(dichloromethyl)dimethylsilane (1b): Bp 70 °C (20 Torr); IR (CH₂Cl₂) 2930, 2856, 1465, 1365, 1264, 832, 785, 740, 701 cm⁻¹; 1 H NMR (CDCl₃) δ 0.21 (s, 6H), 1.00 (s, 9H), 5.41 (s, 1H); 13 C NMR (CDCl₃) δ -7.95, 17.42, 26.97, 62.27. Analytically pure sample could not be obtained because of its sublimation character.

General Procedure for the Reaction of *tert*-Butyldimethylsilyldibromomethyllithium (2a) with aldehydes. An ethereal solution (2 ml) of *tert*-butyl(dibromomethyl)dimethylsilane (0.29 g, 1.0 mmol) was added to a solution of lithium diisopropylamide (1.2 mmol) in Et₂O (3 ml) at -78 °C under argon atmosphere. After being stirred for 1 h at -78 °C benzaldehyde (0.13 g, 1.2 mmol) in Et₂O (1 ml) was added and the reaction mixture was stirred for 20 min at -78 °C. The mixture was quenched with methanol (1 ml). Extractive workup (IM HCl and hexane) followed by purification by silica-gel column chromatography gave 1,1-dibromo-2-(*tert*-butyldimethylsiloxy)-2-phenylethane (6) in 87% yield: Bp 90 °C (1.0 Torr); IR (neat) 2926, 2852, 1455, 1362, 1255, 1135, 1094, 857, 836, 778, 699 cm⁻¹; ¹H NMR (CDCl₃) δ -0.13 (s, 3H), 0.15 (s. 3H), 0.91 (s, 9H), 4.94 (d, J = 5.3 Hz, 1H), 5.63 (d, J = 5.3 Hz, 1H), 7.30–7.45 (m, 5H); ¹³C NMR (CDCl₃) δ -4.94, -4.68, 18.25, 25.69, 51.56, 79.90, 127.48, 128.07, 128.61, 139.75. Found: C, 42.78; H, 5.79%. Calcd for C₁₄H₂₂Br₂OSi: C, 42.65; H, 5.62%.

- **2,2-Dibromo-2-**(*tert*-butyldimethylsilyl)-1-phenylethanol (7): Bp 110 °C (0.5 Torr); IR (neat) 3546, 3448, 2956, 2854, 1464, 1365, 1250, 1027, 821, 712 cm $^{-1}$; 1 H NMR (CDCl $_{3}$) δ 0.35 (s, 3H), 0.37 (s, 3H), 1.15 (s, 9H), 2.65 (d, J = 6.4 Hz, 1H), 5.06 (d, J = 6.4 Hz, 1H), 7.35–7.65 (m, 5H); 13 C NMR (CDCl $_{3}$) δ -3.73, -3.67, 19.91, 28.69, 72.18, 80.10, 127.32, 128.74, 129.26, 138.84. Found: C, 42.77; H, 5.49%. Calcd for C $_{14}$ H $_{22}$ Br $_{2}$ OSi: C, 42.65; H, 5.62%.
- **2,2-Dibromo-1-**(*tert*-butyldimethylsiloxy)-1-phenylpropane (10a): A THF (2 ml) solution of 2,2-dibromo-1-(*tert*-butyldimethylsiloxy)-1-phenylethane (6, 0.39 g, 1.0 mmol) was added to a solution of lithium diisopropylamide (1.2 mmol) in THF (3 ml) at -78 °C. After being stirred for 15 min at -78 °C, methyl iodide (0.09 ml, 1.5 mmol) in THF (1 ml) was added and the reaction mixture was stirred for 1 h at -78 °C. Extractive workup followed by purification by silica-gel column chromatography gave a title compound 10a (0.40 g) in 97% yield: Bp 90 °C (1.0 torr); IR (neat) 2926, 2854, 1454, 1373, 1255, 1099, 1071, 858, 777, 700 cm⁻¹; ¹H NMR (CDCl₃) δ -0.26 (s, 3H), 0.15 (s, 3H), 0.91 (s, 9H), 2.40 (s, 3H), 4.92 (s, 1H), 7.30–7.55 (m, 5H); ¹³C NMR (CDCl₃) δ -5.12, -4.65, 18.20, 25.73, 35.60, 72.78, 83.80, 127.40, 128.48, 129.23, 138.64. Found: C, 43.90; H, 6.02%. Calcd for C₁₅H₂₄Br₂OSi: C, 44.13; H, 5.93%.
- **4.4-Dibromo-5-(***tert***-butyldimethylsiloxy)-5-phenyl-1-pentene (10b):** Bp 105 °C (1.0 Torr); IR (neat) 3078, 3028, 2926, 2854, 1643, 1455, 1361, 1257, 1098, 923, 855, 777, 700 cm⁻¹; 1 H NMR (CDCl₃) 0 0 -0.29 (s, 3H), 0.14 (s, 3H), 0.92 (s, 9H), 3.00 (ddt, J = 15.0, 6.6, 1.3 Hz, 1H), 3.08 (ddt, J = 15.0, 6.6, 1.3 Hz, 1H), 4.98 (s, 1H), 5.20 (ddt, J = 16.8, 1.7, 1.3 Hz, 1H), 5.29 (ddt, J = 10.2, 1.7, 1.3 Hz, 1H), 6.08 (ddt, J

= 16.8, 10.2, 6.6 Hz, 1H), 7.30-7.60 (m, 5H); 13 C NMR (CDCl₃) δ –5.06, –4.57, 18.20, 25.75, 48.50, 79.13, 83.29, 119.65, 127.39, 128.55, 129.58, 133.93, 138.59. Found: C, 46.73; H, 6.01%. Calcd for $C_{17}H_{26}Br_{2}OSi$: C, 47.02; H, 6.03.

(IR*,3S*)-2,2-Dibromo-1,3-diphenyl-3-(*tert*-butyldimethylsiloxy)propanol (5'): Mp 120–121 °C; IR (CH₂Cl₂) 3542, 3050, 2926, 2854, 1454, 1265, 1113, 863, 838, 732, 701 cm⁻¹; ¹H NMR (CDCl₃) δ –0.29 (s, 3H), 0.17 (s, 3H), 0.96 (s, 9H), 3.10 (d, J = 5.4 Hz, 1H), 4.53 (d, J = 5.4 Hz, 1H), 5.27 (s, 1H), 7.30–7.75 (m, 10H); ¹³C NMR (CDCl₃) δ –4.91, –4.28, 18.27, 25.83, 78.65, 81.35, 85.94, 127.33, 127.59, 128.74, 129.43, 138.61, 138.85. Found: C, 50.24; H, 5.64%. Calcd for C₂₁H₂₈Br₂O₂Si: C, 50.41; H, 5.64%. The physical and spectra data of 5 and 5' were identical with those of authentic sample: ¹⁸

(IR*,3S*)-2,2-Dibromo-1-phenyl-1-(*tert*-butyldimethylsiloxy)-3-heptanol (10c'): Bp 115 °C (0.5 Torr); IR (neat) 3546, 3446, 3030, 2926, 2859, 1459, 1362, 1253, 1120, 838, 777, 699 cm⁻¹; ¹H NMR (CDCl₃) δ –0.31 (s, 3H), 0.13 (s, 3H), 0.89 (t, J = 7.0 Hz, 3H), 0.95 (s, 9H), 1.20–2.20 (m, 7H), 3.23 (t, J = 10.0 Hz, 1H), 5.19 (s, 1H), 7.30–7.65 (m, 5H); ¹³C NMR (CDCl₃) δ –5.04, –4.39, 14.03, 18.22, 22.58, 25.78, 28.10, 34.65, 76.79, 80.19, 90.42, 127.48, 128.46, 129.26, 138.56. Found: C, 47.68; H, 6.72%. Calcd for C₁₉H₃₂Br₂O₂Si: C, 47.51; H, 6.71%. The assignment of the stereochemistry of 10c and 10c' were performed by NOE experiment.

1-(*tert*-Butyldimethylsiloxy)-2,2-dibromo-2-(1-hydroxycyclohexyl)-1-phenylpropane (10d): Mp 100–101 °C; IR (CH₂Cl₂) 3474, 2930, 2856, 1452, 1265, 1051, 858, 837, 738, 701 cm $^{-1}$; 1 H NMR (CDCl₃) δ –0.46 (s, 3H), 0.09 (s, 3H), 0.92 (s, 9H), 1.10–2.30 (m, 10H), 3,97 (bs. 1H), 5.28 (s, 1H), 7.30–7.80 (m, 5H); 13 C NMR (CDCl₃) δ –4.74, –4.05, 18.04, 21.76, 22.24, 25.55, 25.81, 31.50, 35.50, 79.56, 80.98, 126.90, 128.88, 131.07, 138.96. Found: C, 48.84; H, 6.81%. Calcd for C₂₀H₃₂Br₂O₂Si: C, 48.79; H, 6.55%.

General Procedure for One-pot Synthesis of 12 (RCH(OSiMe₂-t-Bu)CCl₂E') from 1b. A THF (2 ml) solution of tert-butyl(dichloromethyl)dimethylsilane (1b, 0.24 g, 1.2 mmol) was added to a solution of lithium diisopropylamide (1.4 mmol) in THF (3 ml) at -78 °C under argon atmosphere. After being stirred for 1 h at -78 °C, benzaldehyde (0.11 g, 1.0 mmol) in THF (1 ml) was added and the reaction mixture was stirred for 20 min at -78 °C. Methyl iodide (1.5 mmol) in THF (1 ml) and HMPA (0.24 ml, 1.4 mmol) in THF (1 ml) were added successively to the reaction mixture and the resulting mixture was allowed to warm to room temperature over 5 h. Extractive workup (1M HCl and hexane) followed by purification by silica-gel column chromatography gave 1-(tert-butyldimethylsiloxy)-2,2-dichloro-1-phenylpropane 12a (0.23 g) in 71% yield. When aldehydes were used as second electrophiles, the reaction mixture was allowed to warm to -20 °C and kept there for 1 h before workup. 12a: Bp 90 °C (1.0 Torr); IR (neat) 2928, 2884, 2854, 1455, 1375, 1254, 1105, 1076, 861, 836, 777, 699 cm⁻¹; H NMR (CDCl₃) δ -0.21 (s, 3H), 0.12 (s, 3H), 0.90 (s, 9H), 2.04 (s, 3H), 4.92 (s, 1H), 7.30-7.60 (m, 5H); ¹³C NMR (CDCl₃) δ -5.21, -4.75, 18.16, 25.67, 31.96, 82.83, 92.10, 127.44, 128.42, 129.01, 138.58. Found: C, 56.32; H, 7.65%. Calcd for C₁₅H₂₄Cl₂OSi: C, 56.42; H, 7.58%.

- 5-(*tert*-Butyldimethylsiloxy)-4,4-dichloro-5-phenyl-1-pentene (12b): Bp 95 °C (1.0 Torr); IR (neat) 3080, 2950, 2854, 1644, 1455, 1254, 1105, 930, 858, 837, 777, 699 cm $^{-1}$; ¹H NMR (CDCl $_3$) δ -0.24 (s, 3H), 0.11 (s, 3H), 0.90 (s, 9H), 2.83 (dd, J = 14.7, 6.7 Hz, 1H), 2.97 (dd, J = 14.7, 6.7 Hz, 1H), 4.96 (s, 1H), 5.19 (dd, J = 17.1, 1.4 Hz, 1H), 5.27 (dd, J = 10.2, 1.4 Hz, 1H), 6.05 (ddt, J = 17.1, 10.2, 6.7 Hz, 1H), 7.30–7.40 (m, 3H), 7.45–7.55 (m, 2H); ¹³C NMR (CDCl $_3$) δ -5.15, -4.66, 18.16, 25.69, 46.31, 82.49, 95.04, 120.03, 127.44, 128.49, 129.32, 131.80, 138.32. Found: C, 59.32; H, 7.60%. Calcd for C $_{17}$ H $_{26}$ Cl $_2$ OSi: C, 59.12; H, 7.59%.
- (E)-3-(tert-Butyldimethylsiloxy)-4,4-dichloro-1-phenyl-1-pentene (12c): Bp 100 °C (1.0 Torr); IR (neat) 3010, 2928, 2854, 1650, 1460, 1253, 1130, 1073, 968, 873, 836, 777, 748, 691 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.08 (s, 3H), 0.16 (s, 3H), 0.94 (s, 9H), 2.08 (s, 3H), 4.49 (d, J = 6.8 Hz, 1H), 6.31 (dd, J = 15.9, 6.8 Hz, 1H), 6.68 (d, J = 15.9 Hz, 1H), 7.27–7.45 (m, 5H); 13 C NMR (CDCl₃) δ -4.84, -3.99, 18.22, 25.77, 32.19, 81.64, 91.88, 126.70, 128.09, 128.65, 134.42, 136.23. Found: C, 59.36; H, 7.88%. Calcd for C₁₇H₂₆Cl₂OSi: C, 59.12; H, 7.59%.
- **5-(tert-Butyldimethylsiloxy)-4,4-dichloro-1-octene (12d):** Bp 65 °C (1.0 Torr); IR (neat) 2956, 2856, 1464, 1362, 1257, 1146, 1104, 924, 835, 775 cm $^{-1}$; 1 H NMR (CDCl $_{3}$) δ 0.12 (s, 3H), 0.15 (s, 3H), 0.92 (s, 9H), 0.94 (t, J = 7.3 Hz, 3H), 1.20–2.05 (m, 4H), 2.80–3.00 (m, 2H), 3.90 (dd, J = 7.0, 2.6 Hz, 1H), 5.22 (dq, J = 17.0, 1.7 Hz, 1H), 5.27 (dq, J = 10.2, 1.7 Hz, 1H), 6.03 (ddt, J = 17.0, 10.2, 6.8 Hz, 1H); 13 C NMR (CDCl $_{3}$) δ –3.98, –3.60, 14.17, 18.43, 26.02, 35.95, 46.75, 80.81, 96.27, 119.86, 131.92. Found: C, 53.83; H, 9.29%. Calcd for C $_{14}$ H $_{28}$ Cl $_{2}$ OSi: C, 54.01; H, 9.06%.
- (IR*,3R*)-2,2-Dichloro-1-(4-methoxyphenyl)-3-phenyl-1,3-propanediol: 1,3-Diol monosilyl ether 12e was converted into diol with saturated aqueous KF in the presence of a catalytic amount of n-Bu₄NF in THF and two diastereomers of diol were separated by silica-gel column chromatography. IR (neat) 3382, 2954, 2930, 1710, 1611, 1513, 1250, 1177, 1066, 1032, 832, 731, 700 cm⁻¹; 1 H NMR (CDCl₃) δ 3.04 (bs, 1H), 3.12 (bs, 1H), 3.82 (s, 3H), 5.05 (s, 1H), 5.07 (s, 1H), 6.90 (d, J = 8.8 Hz, 2H), 7.35–7.60 (m, 7H); 13 C NMR (CDCl₃) δ 55.22, 78.92, 79.23, 98.49, 113.20, 127.71, 127.80, 128.80, 128.90, 129.20, 129.93, 130.15, 137.15, 154.97. Found: C, 58.87; H, 4.94%. Calcd for C₁₆H₁₆O₃Cl₂: C, 58.73; H, 4.93%.
- (IR*,3S*)-2,2-Dichloro-1-(4-methoxyphenyl)-3-phenyl-1,3-propanediol: Bp 110 °C (0.5 Torr); IR (neat) 3388, 2954, 2930, 1707, 1611, 1514, 1252, 1178, 1035, 860, 829, 730, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 3.57 (bs, 1H), 3.69 (bs, 1H), 3.82 (s, 3H), 5.27 (s, 1H), 5.30 (s, 1H), 6.91 (d, J=8.9 Hz, 2H), 7.35–7.60 (m, 7H); ¹³C NMR (CDCl₃) δ 55.21, 78.87, 79.17, 95.02, 113.10, 127.68, 127.78, 128.80, 128.87, 129.08, 130.04, 130.23, 137.02, 159.85. Found: C, 58.50; H, 4.96%. Calcd for C $_{16}H_{16}O_{3}Cl_{2}$: C, 58.73; H, 4.93%.
- (IR*,3R*)-2,2-Dichloro-1-phenyl-1,3-heptanediol: Bp 90 °C (0.2 Torr); IR (neat) 3838, 3820, 2956, 2860, 1492, 1455, 1379, 1191, 1089, 1063, 859, 702 cm⁻¹; ¹H NMR (CDCl₃) δ 0.90 (t, J=7.3 Hz, 3H), 1.20–2.15 (m, 6H), 2.26 (d, J=9.3 Hz, 1H), 3.38 (d, J=3.5 Hz, 1H), 3.66 (dt, J=9.3, 1.9 Hz, 1H), 5.31 (d, J=3.5 Hz, 1H), 7.35–7.65 (m, 5H); ¹³C NMR (CDCl₃) δ 13.94, 22.46, 27.82, 32.15, 77.52, 79.50, 100.33, 127.74, 127.84, 128.70, 136.91. Found: C, 56.05; H, 6.58%. Calcd for C $_{13}$ H₁₈Cl₂O₂: C, 56.33; H,6.55%.
- (*E*)-(*IR**,3*R**)-2,2-Dichloro-1,5-diphenyl-4-pentene-1,3-diol: IR (nujol) 3458, 1455, 1198, 1118, 1064, 1046, 966, 906, 835, 737, 700 cm $^{-1}$; ¹H NMR (CDCl₃) δ 2.57 (d, J = 6.5 Hz, 1H), 3.07 (d, J = 4.3 Hz, 1H), 4.47 (t, J = 6.5 Hz, 1H), 5.34 (d, J = 4.3 Hz, 1H), 6.50 (dd, J = 16.0, 6.5 Hz, 1H), 6.74 (d, J = 16.0 Hz, 1H), 7.25–7.65 (m, 10H); ¹³C NMR (CDCl₃) δ 78.01, 79.10, 98.83, 124.98, 126.69, 126.86, 126.95,

127.86, 128.06, 128.28, 128.55, 128.73, 129.04, 135.47, 135.81, 136.76. Found: C, 62.93; H, 5.06%. Calcd for $C_{17}H_{16}Cl_{2}O_{2}$: C, 63.17; H,4.99%.

(*E*)-(*1R**,3*S**)-2,2-Dichloro-1,5-diphenyl-4-pentene-1,3-diol: IR (neat) 3306, 3028, 2920, 1719, 1638, 1493, 1452, 1201, 1123, 1044, 966, 909, 866, 746, 696 cm $^{-1}$; 1 H NMR (CDCl₃) δ 2.97 (bs, 1H), 3.27 (bs, 1H), 4.88 (bd, J = 6.0 Hz, 1H), 5.33 (s, 1H), 6.53 (dd, J = 15.9, 6.0 Hz, 1H), 6.83 (d, J = 15.9 Hz, 1H), 7.25–7.65 (m, 10H); 13 C NMR (CDCl₃) δ 77.63, 78.65, 95.85, 124.69, 126.76, 126.86, 127.70, 127.84, 128.26, 128.34, 128.59, 128.70, 128.81, 129.03, 135.32, 135.95, 136.93. Found: C, 62.88; H, 4.98%. Calcd for C₁₇H₁₆Cl₂O₂: C, 63.17; H, 4.99%.

Reduction of Dichloride 12a with n-Bu₃SnH-Et₃B. A hexane solution of Et₃B (0.96 M, 0.73 ml, 0.7 mmol) was added to a solution of 12a (186 mg, 0.6 mmol) and n-Bu₃SnH (0.47 ml, 1.75 mmol) in hexane (5 ml). The mixture was heated at 80 °C for 24 h. The resulting mixture was concentrated in vacuo and the residual oil was diluted with dichloromethane (20 ml). Potassium fluoride (1.0 g) and saturated aqueous potassium fluoride (1.0 ml) were added and the resulting mixture was stirred at 25 °C for 15 h. The reaction mixture was filtered and filtrate was concentrated. Purification of the residual oil by silica-gel column chromatography gave 1-phenyl-1-tert-butyldimethylsiloxypropane (0.15 g) in 97% yield.

3,3-Dibromo-3-(*tert*-butyldimethylsilyl)-1-phenyl-1-propanol (16a): IR (neat) 3562, 3426, 2958, 2928, 2884, 2856, 1465, 1253, 1039, 835, 820, 776, 761, 698, 668 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.30 (s, 3H), 0.32 (s, 3H), 1.06 (s, 9H), 2.80 (dd, J = 15.3, 2.7 Hz, 1H), 2.87 (dd, J = 15.3, 6.3 Hz, 1H), 2.98 (d, J = 2.4 Hz, 1H), 5.54 (ddd, J = 6.3, 2.7, 2.4 Hz, 1H), 7.25 $^{-}$ 7.50 (m, 5H); 13 C NMR (CDCl₃) δ -5.87, 19.59, 28.46, 55.05, 67.89, 73.70, 125.80, 127.55, 128.68, 144.30. Found: C, 44.13; H, 5.93%. Calcd for C₁₅H₂₄OBr₂Si: C, 44.29; H, 5.95%.

4,4-Dibromo-4-(*tert*-butyldimethylsilyl)-2-butanol (16b): Bp 100 °C (1 Torr); IR (neat) 3390, 2960, 2930, 2896, 2858, 1465, 1366, 1253, 1073, 930, 836, 776, 668 cm $^{-1}$: 1 H NMR (CDCl₃) δ 0.307 (s, 3H), 0.314 (s, 3H), 1.081 (s, 9H), 1.31 (d, J = 6.3 Hz, 3H), 2.53 (dd, J = 15.3, 2.7 Hz, 1H), 2.60 (dd, J = 15.3, 5.7 Hz, 1H), 2.64 (d, J = 2.7 Hz, 1H), 4.59 (m, 1H); 13 C NMR (CDCl₃) δ -5.95, 19.54, 24.06, 28.43, 53.90, 68.11, 68.59. Found: C, 34.67; H, 6.53%. Calcd for C $_{10}$ H₂₂Br $_{2}$ OSi: C, 34.70; H, 6.41%.

4,4-Dibromo-4-(*tert*-butyldimethylsilyl)-1-methoxy-2-butanol (16c): Bp 95 °C (0.5 Torr); lR (neat) 3426, 2928, 2884, 2856, 1465, 1253, 1195, 1126, 1086, 934, 835, 775, 667 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.32 (s, 6H), 1.08 (s, 9H), 2.59 (dd, J = 15.5, 4.8 Hz, 1H), 2.65 (dd, J = 15.5, 3.6 Hz, 1H), 2.74 (d, J = 3.3 Hz, 1H), 3.438 (s, 3H), 3.442 (dd, J = 9.6, 7.2 Hz, 1H), 3.57 (dd, J = 9.6, 3.9 Hz, 1H), 4.54 (m, 1H); 13 C NMR (CDCl₃) δ –5.93, –5.83, 19.53, 28.44, 49.35, 59.09, 67.60, 70.40, 76.55. Found: C, 35.36; H, 6.58%. Calcd for C₁₁H₂₄Br₂O₂Si: C, 35.12; H, 6.43%.

General Procedure for the Reaction of Silyldihalomethyllithium 2 with Oxirane. A reaction of tert-butyldimethylsilyldibromomethyllithium (2a) with styrene oxide is representative. A THF (2 ml) solution of tert-butyl(dibromomethyl)dimethylsilane (1a, 0.29 g, 1.0 mmol) was added to a solution of lithium diisopropylamide (1.2 mmol) in THF (3 ml) at -78 °C. After being stirred for 1 h at -78 °C, styrene oxide (0.14 g, 1.2 mmol) in THF (1 ml) was added and the mixture was warmed to -40 °C over 1 h. The resulting mixture was quenched with methanol and stirred another 10 min at room temperature. Extractive workup (1M HCl and hexane) followed by purification by silica-gel column chromatography gave 1,1-dibromo-3-(tert-butyldimethylsiloxy)-3-phenylpropane (17c, 0.27 g) in 65 % yield: Bp 135 °C (1.0 Torr); IR (neat) 2948, 2928, 2884, 2854, 1471, 1456, 1362, 1255, 1156, 1089, 1002, 929, 837, 777, 699, 615 cm⁻¹; ¹H NMR (CDCl₃) δ -0.24 (s, 3H), 0.05 (s, 3H), 0.86 (s, 9H), 2.55 (ddd, J = 14.5, 9.2, 4.0 Hz, 1H), 2.79 (ddd, J = 14.5, 9.2, 4.0 Hz, 1H), 4.87 (dd, J = 9.2, 3.5 Hz, 1H), 5.69 (dd, J = 9.8, 4.0 Hz, 1H), 7.20–7.35 (m, 5H); ¹³C NMR (CDCl₃) δ -4.96, -4.47, 18.08, 25.79, 42.91, 56.18, 73.62, 126.07, 127.83, 128.44, 143.14. Found: C, 44.00; H, 5.94%. Calcd for C $_1$ 5H₂₄Br₂OSi: C, 44.13; H, 5.93%.

- 1,1-Dibromo-3-(*tert*-butyldimethylsiloxy)butane (17a): Bp 90 °C (1.0 Torr); IR (neat) 2952, 2926, 2886, 2884, 1463, 1375, 1256, 1135, 1046, 967, 836, 775, 683 cm $^{-1}$; 1 H NMR (CDCl₃) 8 0.08 (s, 3H), 0.11 (s, 3H), 0.89 (s, 9H), 1.18 (d, J=6.0 Hz, 3H), 2.41 (ddd, J=14.4, 10.2, 3.0 Hz, 1H), 2.54 (ddd, J=14.4, 9.0, 3.6 Hz, 1H), 4.00 (ddq, J=3.0, 9.0, 6.0 Hz, 1H), 5.72 (dd, J=10.2, 3.6 Hz, 1H); 13 C NMR (CDCl₃) 8 C-4.90, -4.27, 17.84, 23.33, 25.73, 43.38, 55.06, 67.04. Found: C, 34.99; H, 6.56%. Calcd for C₁₀H₂₂Br₂OSi: C, 34.70; H, 6.41%.
- **1,1-Dibromo-3-(***tert*-butyldimethylsiloxy)propane (17b): Bp 80 °C (1 Torr); IR (neat) 2952, 2926, 2856, 1471, 1387, 1256, 1161, 1104, 932, 836, 777, 686 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.07 (s, 6H), 0.90 (s, 9H), 2.58 (dt, J = 6.6, 5.7 Hz, 2H), 3.72 (t, J = 5.7 Hz, 2H), 5.84 (t, J = 6.6 Hz, 1H); 13 C NMR (CDCl₃) δ -5.60, 18.14, 25.76, 43.16, 48.13, 60.70. Found: C, 32.82; H, 6.01%. Calcd for C₉H₂₀Br₂OSi: C, 32.55; H, 6.07%.
- **5,5-Dibromo-3-**(*tert*-butyldimethylsiloxy)-1-pentene (17d): Bp 90 °C (1 Torr); IR (neat) 3078, 3008, 2952, 2928, 2884, 2856, 1645, 1463, 1419, 1362, 1253, 1086, 923, 836, 776, 680, 562 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.05 (s, 3H), 0.10 (s, 3H), 0.90 (s, 9H), 2.46 (ddd, J = 14.4, 9.3, 3.9 Hz, 1H), 2.61 (ddd, J = 14.4, 9.0, 4.5 Hz, 1H), 4.25 (m, 1H), 5.13 (d, J = 10.2 Hz, 1H), 5.24 (d, J = 17.1 Hz, 1H), 5.69 (dd, J = 9.3,4.5 Hz, 1H), 5.78 (ddd, J = 17.1, 10.2, 7.2 Hz, 1H); 13 C NMR (CDCl₃) δ -4.96, -4.20, 17.98, 25.74, 42.42, 63.42, 72.68, 115.94, 139.83. Found: C, 36.91; H, 6.21%. Calcd for C₁₁H₂₂Br₂OSi: C, 36.89; H, 6.19%.
- **3-(tert-butyldimethylsiloxy)-1,1-dichlorobutane** (17e): Bp 110 °C (8 Torr); IR (neat) 2954, 2928, 2888, 2856, 1472, 1363, 1257, 1139, 1050, 973, 836, 775, 754, 665 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.89 (s, 9H), 1.18 (d, J = 6.0 Hz, 3H), 2.21 (ddd, J = 14.0, 9.6, 3.0 Hz, 1H), 2.32 (ddd, J = 14.0, 9.3, 3.6 Hz, 1H), 4.03 (ddq, J = 9.3, 3.0, 6.0 Hz, 1H), 5.81 (dd, J = 9.6, 3.6 Hz, 1H); 13 C NMR (CDCl₃) δ -5.09, -4.33, 17.82, 23.57, 25.69, 53.25, 65.61, 71.29. Found: C, 46.58; H, 8.84%. Calcd for C₁₀H₂₂Cl₂OSi: C, 46.69; H, 8.62%.
- **3-(tert-butyldimethylsiloxy)-1,1-dichloropropane (17f):** Bp 100 °C (9 Torr); IR (neat) 2952, 2928, 2880, 2856, 1472, 1387, 1257, 1108, 938, 835, 777, 756, 664 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.06 (s, 6H), 0.89 (s, 9H), 2.38 (dt, J = 6.3, 5.4 Hz, 2H), 3.78 (t, J = 5.4 Hz, 2H), 5.92 (t, J = 6.3 Hz, 1H); 13 C NMR (CDCl₃) δ -5.66, 18.13, 25.74, 46.40, 59.19, 71.07. Found: C, 44.53; H, 8.54%. Calcd for C₉H₂₀Cl₂OSi: C, 44.44; H, 8.29%.
- 1-(tert-Butyldimethylsiloxy)-3,3-dichloro-1-phenylpropane (17g): Bp 100 °C (1.0 Torr); IR (neat) 2952, 2928, 2886, 2854, 1464, 1363, 1254, 1093, 1005, 937, 836, 777, 745, 698, 671, 611 cm⁻¹; $^{1}\mathrm{H}$ NMR (CDCl3) δ –0.22 (s, 3H), 0.05 (s, 3H), 0.88 (s, 9H), 2.37 (ddd, J = 14.1, 9.3, 3.3 Hz, 1H), 2.60 (ddd, J = 14.1, 9.6, 3.9 Hz, 1H), 4.84 (dd, J = 9.3, 3.3 Hz, 1H), 5.82 (dd, J = 9.6, 3.6 Hz, 1H), 7.20–7.40 (m, 5H); $^{13}\mathrm{C}$ NMR (CDCl3) δ –5.29, –4.68, 17.67, 25.68, 54.41, 70.91, 72.33, 126.09, 127.91, 128.54, 143.46. Found: C. 56.19; H, 7.71%. Calcd for C $_{15}\mathrm{H}_{24}\mathrm{Cl}_{2}\mathrm{OSi}$: C, 56.42; H, 7.57%.
- **2-(***tert*-Butyldimethylsiloxy)-1,4,4-trichlorobutane (17h): Bp 70 °C (0.5 Torr); IR (neat) 2952, 2928, 2886, 2856, 1465, 1390, 1363, 1257, 1153, 1092, 935, 836, 776, 665 cm $^{-1}$; 1 H NMR (CDCl₃) 1 8 0.13 (s, 6H), 0.91 (s, 9H), 2.44 (ddd, J = 14.4, 8.1, 4.2 Hz, 1H), 2.52 (ddd, J = 14.4, 9.3, 3.3 Hz, 1H), 3.44 (dd, J = 11.4, 6.3 Hz, 1H), 3.51 (dd, J = 11.4, 3.9 Hz, 1H), 4.10 (m, 1H), 5.81 (dd, J = 9.3, 4.2 Hz, 1H); 13 C NMR (CDCl₃) 1 8 $^{-4}$ 97, $^{-4}$ 57, 17.87, 25.60, 47.72, 48.82, 69.36, 70.55. Found: C, 41.38; H, 7.39%. Calcd for C₁₀H₂₁Cl₃OSi: C, 41.17; H, 7.26%.
- General Procedure for One-pot synthesis of 19 (RCH(OSiMe₂-t-Bu)CH₂CX₂E') from 1. A THF (2 ml) solution of *tert*-butyl(dichloromethyl)dimethylsilane (0.20 g, 1.0 mmol) was added to a solution of lithium diisopropyl amide (1.2 mmol) in THF (3 ml) at -78 °C. After being stirred for 1 h at -78 °C, propylene oxide (0.07 g, 1.2 mmol) in THF (1 ml) was added and the mixture was warmed to -40 °C over 1 h. The resulting mixture was cooled to -78 °C and iodomethane (0.21 g, 1.5 mmol) and HMPA (0.24 ml, 1.4 mmol) in THF (1 ml) were added successively. The whole mixture was allowed to warm to room temperature

- over 5 h. Extractive workup (1*M* HCl and hexane) followed by purification by silica-gel column chromatography gave 2-(*tert*-butyldimethylsiloxy)-4,4-dichloropentane (19h, 0.16 g) in 68% yield. When aldehydes were used as second electrophiles, the reaction mixture was allowed to warm to -20 °C and kept there for 1 h before workup. 19h: Bp 105 °C (9 Torr); IR (neat) 2954, 2928, 2894, 2856, 1464, 1377, 1257, 1138, 1037, 976, 938, 836, 775, 698, 655, 599 cm⁻¹; 1 H NMR (CDCl₃) δ 0.09 (s, 6H), 0.89 (s, 9H), 1.26 (d, J = 6.0 Hz, 3H), 2.20 (s, 3H), 2.38 (dd, J = 14.7, 3.9 Hz, 1H), 2.46 (dd, J = 14.7, 6.3 Hz, 1H), 4.25 (ddq, J = 6.3, 3.9, 6.0 Hz 1H); 13 C NMR (CDCl₃) δ -4.61, -4.05, 17.81, 25.05, 25.79, 38.05, 58.72, 66.66, 89.43. Found: C, 48.44; H, 9.10%. Calcd for C₁₁H₂₄Cl₂OSi: C, 48.70; H, 8.92%.
- **2,2-Dibromo-4-**(*tert*-butyldimethylsiloxy)pentane (19a): Bp 95 °C (1 Torr); IR (neat) 2954, 2928, 2892, 2854, 1463, 1376, 1257, 1136, 1098, 1033, 972, 836, 774, 652 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.10 (s, 3H), 0.11 (s, 3H), 0.89 (s, 9H), 1.27 (d, J = 6.0 Hz, 3H), 2.59 (s, 3H), 2.62 (d, J = 5.1 Hz, 2H), 4.23 (dq, J = 5.1, 6.3 Hz, 1H); 13 C NMR (CDCl₃) δ -4.43, -3.92, 17.80, 24.91, 25.82, 41.82, 61.78, 66.70, 68.50. Found: C, 36.85; H, 6.81%. Calcd for C₁₁H₂₄Br₂OSi: C, 36.68; H, 6.72%.
- 4-(tert-butyldimethylsiloxy)-2,2-dichloro-1-phenyl-1-pentanol (19c, 53:47 diastereomeric mixture): Bp 125 °C (0.3 Torr); IR (neat) 3424, 2952, 2926, 2892, 2854, 1456, 1377, 1256, 1128, 1052, 1004, 966, 833, 774, 700, 604 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.12 (s, 1.59H), 0.18 (s, 1.41H), 0.90 (s, 4.77H), 0.91 (s, 4.23H), 1.29 (d, J = 6.3 Hz, 1.59H), 1.31 (d, J = 6.3 Hz, 1.41H), 2.34 (dd, J = 15.3, 4.2 Hz, 0.53H), 2.43 (dd, J = 15.3, 6.6 Hz, 0.53H), 2.44 (dd, J = 15.3, 5.4 Hz, 0.47H), 2.86 (dd, J = 15.3, 8.4 Hz, 0.47H), 3.50 (d, J = 3.6 Hz, 0.53H), 4.07 (d, J = 4.5 Hz, 0.47H), 4.41 (m, 1H), 5.12 (d, J = 4.5 Hz, 0.47H), 5.14 (d, J = 3.6 Hz, 0.53H), 7.30–7.40 (m, 3H), 7.50–7.65 (m, 2H); 13 C NMR (CDCl₃) δ –4.49, –4.36, –4.30, –4.02, 17.92, 24.69, 24.90, 25.83, 52.28, 54.22, 66.91, 66.97, 79.28, 81.16, 94.55, 96.12, 143.91, 144.18, 144.90, 145.16, 145.64, 145.68, 153.43, 153.55. Found: C, 56.10; H, 7.82%. Calcd for C $_{17}$ H₂₈Cl₂O₂Si: C, 56.19; H, 7.77%.
- 1-(*tert*-Butyldimethylsiloxy)-3,3-dichlorobutane (19d): Bp 110 °C (8 Torr); IR (neat) 2952. 2928, 2882, 2856, 1472, 1382, 1257, 1110, 902, 838, 776, 699 cm⁻¹; 1 H NMR (CDCl₃) δ 0.08 (s, 6H), 0.90 (s, 9H), 2.20 (s, 3H), 2.49 (t, J = 6.6 Hz, 2H), 3.94 (t, J = 6.6 Hz, 2H); 13 C NMR (CDCl₃) δ -5.56, 18.07, 25.76, 37.98, 51.87, 60.05, 88.88. Found: C, 46.67; H, 8.57%. Calcd for C₁₀H₂₂Cl₂OSi; C, 46.69; H, 8.62%.
- **4-(***tert*-butyldimethylsiloxy)-**2,2-dichlorobutanal** (**19e**): Bp 95 °C (4 Torr); IR (neat) 2952, 2928, 2882, 2856, 1751, 1472, 1390, 1363, 1257, 1105, 977, 837, 777, 663, 610 cm $^{-1}$; 1 H NMR (CDCl₃) δ 0.03 (s, 6H), 0.86 (s, 9H), 2.64 (t, J = 5.7 Hz, 2H), 3.83 (t, J = 5.7 Hz, 2H), 9.15 (s, 1H); 13 C NMR (CDCl₃) δ -5.69, 18.20, 25.77, 46.18, 58.93, 87.65, 184.43. Found: C. 44.32; H, 7.73%. Calcd for C $_{10}$ H₂₀Cl₂O₂Si: C, 44.28; H, 7.43%.

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